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Kinetic Monte Carlo Approach to Cluster Growth in Magnetron Plasmas

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Abstract

The control of nanoscale structures is of great importance to the development of technologies. Specifically, the nanoparticle may be considered as the essential building block of a variety of structures. A recent example [1] shows that gold particles can be used as seeds for the growth of nanowires with dimensions and shape controlled by the size of the seed. We present here a kinetic Monte Carlo scheme to simulate the growth [2] and size distributions of metallic clusters formed in a magnetron discharge by the agglomeration of sputtered atoms. Given a set of experimental conditions, we show that the size distribution of emitted cluster can be determined within the growth phase. The use of the "first reaction" [3] method enables us to significantly speed-up the simulation by following only representative cluster sizes within the system. We present some preliminary results and outline the general ideas to simulate a macroscopic system.

Simulation Models

Constant He flow velocity:

 $\bar{v}_g = 2 \times 10^3$ cm/s

(Space average of position-dependent flow velocity: $\bar{v}_g(x) = \frac{v_e R^2}{(R+x\tan\alpha)^2}$; chamber length x = 4cm, nozzle radius R = 3mm, walls angled at $\alpha = 45^{\circ}$, and exit velocity $v_e = 2.6 \times 10^4$ cm/s [4])

Theoretical Approach

• Metal atom velocity after an elastic collision:

 $v_f = v_i(1-\mu) + \mu V_g$

where $\mu = 2m/(M+m)$, $V_g \sim \mathcal{N}[\bar{v}_g, \sigma_g^2]$

• After many collisions (N_c) :

 $v(N_c) \sim \left(e^{bN_c} \mathcal{N}[\bar{v}_0 - \bar{v}_g, \sigma_0^2 - \frac{m}{M}\sigma_g^2] + \mathcal{N}[\bar{v}_g, \frac{m}{M}\sigma_g^2] \right)$

 $b = \ln(1 - \mu)$, σ_0^2 is initial variance of metal atom velocities

• Combine:

1) mean value approximation for $v(N_c)$ to get $v(N_c)$

Growth Model

- 3D treatment of clusters
- Initialize space with cluster nuclei (metal dimers)
- Constant monomer background: $n_m = 10^{13} \text{cm}^3$
- Metal assumed to be thermalized with gas background at 300 K

Growth rate:

$$k(N_m) = k_0 N_m^{2/3} n_m$$

reduced growth constant $k_0 = \sqrt{\frac{8k_BT}{\pi M}} \pi r_m^2$, atoms in cluster N_m

1.5 1.2 $\begin{cases}
12 \\
8 \\
4 \\
7^{r}(t) \\
9^{r}(t) \\
9^{r}($

Kinetic Monte Carlo

Physical processes occur according to predetermined rates
Models are only as accurate as the processes that are included
Start with a simplified base list of essential processes

Primary Processes

Metal-Gas Collisions

- Atom/Cluster Movement
- Cluster Nucleation
- Single Atom Adhesion to a Cluster

Deconstruction to Stages

- I. **Thermalization:** Highly energetic sputtered atoms equilibrate with a buffer gas background
- Nucleation: 3-Body Collisions (1 gas–2 metal atoms) result in bound metal dimers that serve as the nucleus for larger clusters
 Growth: Single metal atoms adhere to a cluster surface

Thermalization Model

Model Details

- 1D model of discrete spatial slices
- Space filling neutral buffer gas background described by: atomic properties (*r_g*, *m*), pressure (*P_g*), temperature (*T_g*), and a constant flow velocity (*v̄_g*)
 Metal atoms followed individually through the system: atomic properties (*r_m*, *M*) and intial energy (*E_m*)

2) the relation $N_c(t) = \int_0^t \nu(t') dt'$ \rightarrow a differential equation for $\nu(t)$, the solution is used in (2) to find:

$$N_{c}(t) = \frac{1}{b} \left\{ \ln \left[\frac{\nu_{f}}{\nu_{f} \cosh(\nu_{f} bt) - \nu_{0} \sinh(\nu_{f} bt)} \right] \right\}$$

 v_f and v_0 are the mean final and initial collision frequencies. • $N_c(t)$ is plugged into the expression for $v(N_c)$ to give:

$$v(t) \sim \frac{\nu_f}{\nu_f \cosh(\nu_f bt) - \nu_0 \sinh(\nu_f bt)} \mathcal{N}[\bar{v}_0 - \bar{v}_g, \sigma_0^2 - \frac{m}{M}\sigma_g^2] + \mathcal{N}[\bar{v}_g, \frac{m}{M}\sigma_g^2]$$

Growth Algorithm



- Single atom growth \rightarrow many MC steps: d = 10nm $\approx 30,000$ atoms
- "First-Reaction" kMC: events are time-ordered via rate-based exponential distributions



FIG. 4: *Size Distribution Evolution:* Several snapshots during the simulated growth process are shown. The mean cluster size (red lines) grows linearly in time. The inset shows the mean size and variance as functions of time along with plots of an analytical solution of the model (dashed lines) for these values.

Th	eoretical Approach	
• $k(N_m) = \frac{dN_m}{dt}$; liquid drop model, $r = r_m N_m^{1/3}$: $\frac{dr}{dt} = \frac{1}{3}k_0 r_m n_m$ • Solution gives the mean radius:		
	$\langle r(t) \rangle = \frac{1}{3}k_0 r_m n_m t + r_2$	

 r_2 is the dimer radius ensuring cluster formation at t = 0.

• Taking the variance of this equation:

$$\operatorname{Var}[r(t)] = \left(\frac{1}{3}k_o r_m n_m\right)^2 \operatorname{Var}[t]$$

• Collision cross section from hard sphere model: $\sigma = \pi (r_m + r_g)^2$ • Movement rate:



• Collision rate:





- Multiple growth steps contained in one MC step by summing time samples
- Central Limit Theorem for large growth events: one MC step, one time sample, many growth steps



FIG. 2: Normal Distribution Reproduction: Various binning schemes for representative cluster radii and their reproduction of normal distributions. While each tested scheme can reproduce a wide distribution within reason, the cubic and exponential schemes fail in representing a narrow distribution.

• Several schemes tested (Fig. 2) of representative cluster sizes (linear, quadratic, cubic, and exponential growth with an index)



• Cumulative time variance to proceed from 2 to N_m atoms from rate-based exponential distributions:

$$Var[t] = \sum_{j=2}^{N_m - 1} \frac{1}{k(j)^2} = \sum_{j=2}^{N_m - 1} \frac{1}{(k_0 j^{2/3} n_m)^2}$$

$$\Rightarrow Var[r(N_m)] = \frac{r_m^2}{9} \sum_{j=2}^{N_m - 1} \frac{1}{j^{4/3}}$$

is a generalized harmonic number: $H_a^{(b)} = \sum_{j=1}^{a} \frac{1}{j^b}$
equation gives: $N_m(t) = \left(\frac{1}{3}k_0 n_m t + 2^{1/3}\right)^3$

• Substitute into the variance expression:



Outlook

• Sum

Rate

- Detailed comparison of models to experiments
- Treatment of nucleation stage to bridge thermalization and growth
- Combine stages into a single consistent simulation
- Expansion of interesting and relevant processes
- Cluster-cluster collisions
- Cluster charging
- Extension to reactive materials

time [×10⁻⁶ s]

FIG. 1: Thermalization of Ag atoms in a He buffer gas Thermalization is quantified via the mean and variance of the metal atom velocity distribution. Theoretical results for the mean velocity are fully consistent with simulations, and the variance is consistent with the equilibrated state.

Example Parameters

• Spatial slices: *l* =0.01cm

• Sputtered Ag energy:

 $E_m = 15 \text{eV}$ corresponding to 300V magnetron discharge [4]

• He buffer temperature:

 $T_g = 300 \text{K} \Rightarrow \text{mean thermal veloicty } v_{tg} = \sqrt{3k_B T_g/m}$

• He buffer pressure:

 $P_g = 100 \text{ Pa} \Rightarrow \text{gas} \text{ number density } n_g \approx 2.4 \times 10^{16} \text{cm}^{-3}$

FIG. 3: Algorithm Comparison: A check of the the growth algorithm comparing the quadratic scheme with the continuous growth process. Gaussian fits over the resulting distributions are nearly identical, but the simplified algorithm is computationally three orders of magnitude faster.

- Test model of cluster growth with quadratic scheme against a full calculation (Fig. 3)
- Comparing Gaussian fits: full $\rightarrow \mu = 11.8212$ and $\sigma = 0.0901856$; quadratic $\rightarrow \mu = 11.8206$ and $\sigma = 0.0871358$
- 3 orders of magnitude improvement in computation time: full \rightarrow 26 hours and 5 \times 10¹⁰ MC steps; quadratic \rightarrow 1.5 minutes and 8 \times 10⁷ MC steps

Summary

 Successful implementation of kMC to model various stages of cluster growth

• Development of a fast growth algorithm to treat large cluster sizes

• Development of theoretical analysis to compliment simulation

References

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